

Quantitative Human Physiology Solution Manual

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Dilution gives $C_1 \times V_1 = C_2 \times V_2$; $0.025 \text{ mL} \times 0.1 \text{ M} = 100 \text{ mL} \times C_{\text{ATP}}$ $C_{\text{ATP}} = 0.025 \text{ mL} \times 0.1 \text{ M} / 100 \text{ mL} = 25 \times 10^{-6} \text{ M}$ $A = \epsilon \times C \times l = 15.4 \times 10^3 \text{ M}^{-1} \times 25 \times 10^{-6} \text{ M} = 0.385$

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Jose Venegas MASSACHUSETTS INSTITUTE OF TECHNOLOGY ... SOLUTIONS February 19, 2004 .

Problem 1 A twenty-year-old student was involved in an automobile accident and sustained a chest injury. PROBLEM SET 2 SOLUTIONS February 19,

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